ENT-KAURENE DITERPENOIDS FROM RABDOSIA ERIOCALYX

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Abstract—Two new ent-kaurene diterpenoids, maoecrystal I and maoecrystal J, were isolated from the leaves of Rabdosia eriocalyx. Their structures were determined by detailed spectroscopic analyses. Both maoecrystal I and J inhibited root growth of lettuce seedlings.

INTRODUCTION

Rabdosia eriocalyx Hara is widely distributed in Yunnan, China. It is used in folk medicine to reduce swellings. The leaves of R. eriocalyx Hara collected in different areas in Yunnan, have yielded a series of new ent-kaurene diterpenoid derivatives [1, 2]. As part of our search for biologically active constituents of that plant, the present paper deals with the isolation and the structure elucidation of two new ent-kaurene diterpenoids which we have named maoecrystal I (1) and J (2).

RESULTS

The formula for both compounds were elucidated from the mass spectral data, elemental analysis and ¹H and ¹³C NMR data of 1 and 2 and their derivatives (See Tables 1 and 2, as well as Experimental Section).

Compound 1, $C_{22}H_{30}O_8$, gave UV, IR and ¹H and ¹³C NMR spectra consistent with the presence of a five-membered ketone conjugated with an exocyclic methylene [240.2 (3.6) nm; 1705 (C=O), 1641 (C=C) cm⁻¹, $\delta_{\rm H}$ 5.93 and 5.47 (each 1H, s), $\delta_{\rm C}$ 116.3 (=CH₂), 153.8 (C=, exo-methylene) and 210.9 (ketone)], which could be represented by partial structure A (Fig. 1). An ester was identified as an acetate by the presence of a three-proton singlets at $\delta_{\rm H}$ 2.04 (s) in addition to the signals at $\delta_{\rm C}$ 170.8 (s) and 20.5 (q). A hemiacetal carbon was evident from carbon signal at $\delta_{\rm C}$ 96.5 (s).

Partial structures B and C (Fig. 1) were apparent from ${}^{1}\text{H}^{-1}\text{H}$ COSY and ${}^{1}\text{H}^{-1}^{3}\text{C}$ COSY spectra. An AX system at δ_{H} 3.89 and 2.02 (each 1H, 5.0 Hz, d) and δ_{C} 73.6 (d) and 52.3 (d) could be assigned as partial structure D, directly. The remaining signals were assigned to two oxygenated methylenes corresponding to partial structure E (Fig. 1). Combination of partial structures A–E by superposition of the identically numbered quaternary carbon and comparison of the spectral data with those of related ent-kaurene structures [3, 4] as well as a consideration of the

Table 1. ¹H NMR data of compounds 1 and 2 (500 MHz, CD₂OD, TMS as int. stand.)

Н	1	2
1	3.53 br d 3.0	1.72 t 3.0
1	_	1.29 br d 14.0
2	1.93 ddd 15.0, 3.0, 3.0	1.56 m
2 3	1.94 ddd 15.0, 3.0, 3.0	1.79 dd 15.0, 3.0
3	3.87 d 3.0	5.06 br s
3		2.01 s
5	2.02 d 5.0	1.90 d 5.0
6	3.89 d 5.0	3.87 d 5.0
9	2.03 ddd 12.5, 6.0, 2.0	1.40 dd 14.0, 5.0
11	1.74 ddd 14.0, 8.0, 2.0	1.73 overlap
11	1.81 ddd 14.0, 13.0, 12.5	1.81 overlap
12	2.35 ddd 13.5, 8.0, 8.0	2.40 ddd 13.5, 8.0, 8.0
12	1.46 ddd 13.5, 13.0, 7.0	1.50 m
13	3.11 dd 7.0, 5.0	3.14 dd 8.0, 5.0
14a	2.20 d 12.0	2.23 d 12.0
14b	2.10 dd 12.0, 5.0	2.14 overlap
17a	5.93 s	5.99 s
17b	5.47 s	5.53 s
Me-18	1.25 s	1.19 s
19a	4.32 d 11.7	4.49 d 11.7
19ь	4.37 d 11.7	4.43 d 11.7
Ac-19	2.04 s	2.11 s
20a	4.05 dd 10.5, 1.0	4.05 br d 10.5
20ь	3.71 dd 10.5, 2.0	3.97 br d 10.5

Assignments are based on ¹H-¹H, ¹³C-¹H COSY and NOESY measurements.

structures of diterpenoids which have been reported so far from the genus of *Rabdosia* [3], confirmed that compound 1 has the basic skeleton: ent- 7α -hydroxy- 7β , 20-epoxykaur-16-en-15-one (3).

Of the ¹H NMR data for the two oxygenated methylenes, the signals at $\delta_{\rm H}$ 4.05 (1H, dd, J = 10.5, 1.0 Hz) and 3.71 (1H, dd, J = 10.5, 2.0 Hz) were assigned to H_a-20 and H_b-20 since a small coupling was observed between H-5

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OH

Αc

Table 2. 13C NMR data of 1 and 2*

C	1†	2
1	66.5 CH-O	22.1 ^b CH ₂
2	31.2 CH ₂	22.6 ^b CH ₂
3	71.4 CH-O	72.7° CH–O
4	41.9a C	41.3 C
5	52.3 CH	57.4 CH
6	73.6 CH-O	73.2° CH-O
7	96.5 O-C-O	96.0 O-C-O
8	59,8 C	60.2 C
9	45.8 CH	49.9 CH
10	43.0° C	36.3 C
11	16.3 CH ₂	16.7 CH ₂
12	29.7 CH ₂	29.4 CH ₂
13	35.2 CH	35.0 CH
14	27.0 CH ₂	26.7 CH ₂
15	210.9 C=O	210.3 C=O
16	153.8 C=	153.7 C=
17	$116.3 \text{ H}_2\text{C} =$	$116.7 \text{ H}_2\text{C} =$
18	22.6 Me	21.7 ^d Me
19	67.4 CH ₂ O	66.5° CH ₂ -O
20	66.2 CH ₂ O	66.4° CH ₂ -O
Ac	170.8	170.8
Ac	_	170.3
Ac	20.5	21.0 ^d
Ac	_	20.6 ^d

^{*}Chemical shifts (δ) in ppm relative to pyridine; Multiplicities of signals were determined by INEPT techniques.

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and $\rm H_a$ -20, and H-9 and $\rm H_b$ -20 due to W-type long-range coupling [4]. The other signals were attributed to an oxygenated methylene at C-4, because only one methyl signal, $\delta_{\rm H}$ 1.25 (3H, s) and $\delta_{\rm C}$ 22.6 (q), was observed. The positions of the acetoxy group was confirmed by the presence in the 500 MHz ¹H NMR spectrum of 1 measured in pyridine- d_5 of signals at $\delta_{\rm H}$ 6.61 (OH, 1H, d, 8.0 Hz), 4.40 (H-6, 1H, dd, 8.0, 5.3 Hz) and 2.70 (C₅-H, 1H, d, 5.3 Hz).

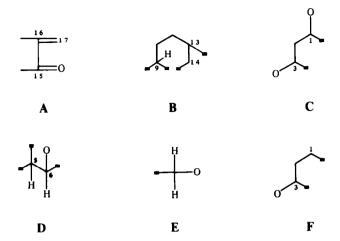
Compound 1 was acetylated with acetic anhydride in pyridine to afford diacetate 4 and monoacetate 5. The hydroxyl groups at C-1 and C-3 in 3 have been acetylated as shown by the downfield shift of $\delta_{\rm H}$ 3.53 and 3.87 to $\delta_{\rm H}$ 4.60 and 5.10, respectively. To sum up the above results, the C-19 hydroxy methyl group should be acetylated in 1.

The unambiguous assignment of all protons of 1 was achieved by a NOESY experiment. Observation of NOEs among H_b -20, H_b -19 and H-3, as well as between H_a -20 and H-1 confirmed that C-19 is axial and that H-1 (δ_H 3.53) and H-3 (δ_H 3.87) are equatorial, respectively. Meanwhile, the coupling constant (6.0 Hz) due to H-5 and H-6 and the NOE effect between 18-Me and H-6, H-6 and H_a -19 suggested that H-6 has equatorial orientation.

The structure of compound 2, $C_{24}H_{32}O_8$, was determined on the basis of the similarity of its INEPT and $^1H^{-1}H$ COSY spectral data to those of 1. Instead of partial structure C in 1, partial structure F was revealed, indicating that 2 lacked a hydroxyl group at C-1 or C-3. The signals in the 1H NMR spectrum (500 MHz, pyridine- d_5) at δ_H 7.17 (OH, 1H, d, 8.0 Hz), 4.40 (H-6, 1H, dd, 8.0, 5.3 Hz) and 2.22 (H-5, 1H, d, 5.3 Hz), showed the presence of a free hydroxyl group at C-6. Furthermore, the fact that C-6 and C-7 have free hydroxyl groups was also supported by periodate oxidation of 2 with periodic

[†]Assignments are based on ¹H-¹H, ¹³C-¹H COSY techniques.

a-eAssignments may be interchanged.



Partial structures A-F

Fig. 1.

acid in methanol to give a spirolactone-type diterpenoid (6) [3]. Consequently, the C-19 hydroxymethyl group and the C-1 or C-3 hydroxyl group had to be acetylated.

The NOEs effect between H_b-20, H_b-19 and H-3 confirmed that H-3 ($\delta_{\rm H}$ 5.06 ppm) is equatorial. Moreover, the NOE effects between H-1 (axial, $\delta_{\rm H}$ 1.72), H-5 and H-9 suggested that they are axial. On the other hand, the NOE effect between H_a-14 and H_a-20 in both of 1 and 2 established the stereostructure to be as depicted in Fig. 2. In order to establish the absolute configurations of 1 and 2, a dihydro compound of 2 (7) was obtained by catalytic hydrogenation of 2. The CD spectrum of 7 could be analysed in a manner applicable to that of a bridgedring (3, 2, 1) structure [5]. The negative Cotton effect indicated that the absolute configuration of the D-ring is the same as that of other similar ent-kaurene diterpenes [6]. The stereochemistry of the new methyl at C-16 in 7 was assigned β -orientation, even if 16R stereochemistry, from its chemical shift (δ_H 1.08, 3H, d, 6 Hz; and 2.45, 1H, quintet, 6 Hz) [6].

Compounds 1 and 2 inhibited root growth of lettuce seedlings with MICS of less than 200 and 20 ppm, respectively. This effect might be due to the α -methylene-cyclopentanone moiety binding to a sulphydryl enzyme.

EXPERIMENTAL

Mps: uncorr. CC: silica gel (Wakogel C-200) and Kieselgel 60 (230-400 mesh, Merck); TLC: Kieselgel 60 F_{254} (Merck); HPLC: ODS C_{18} (6 × 250 mm, detection 240 nm).

Extraction and isolation of diterpenoids. Dried and finely powdered leaves of R. eriocalyx (Dunn) Hara (3.0 kg), collected on Oct. 1985 at Yanzhonghai, Yunnan, China, were extracted with MeOH (3×3 l) at room temp. for 20 days. Filtration and evapn of the solvent yielded 110 g of residue which was dissolved in MeOH-H₂O (1:9) and shaken with 3×2 l of Et₂O. The Et₂O-soln was evapd in vacuo to yield 75 g of residue. This was treated (\times 2) with activated charcoal in MeOH (1.5 l), filtered and the solvent evapd to yield 44 g of a yellow gum which was subjected to CC over silica gel (700 g). The column was eluted successively with hexane–EtOAc (9:1, 4:1, 3:1, 13:9, and 1:1), EtOAc, and EtOAc–MeOH (4:1). Fractions 27–32 gave 4.2 g crude crystal-

1

NOE's observed in maoecrystal I (1) and maoecrystal J (2)

2

Fig. 2.

line 2. Fractions 36-46 (4.52 g) were bulked and subjected to silica gel CC on Kieselgel 60 (230-400 mesh, Merck) to give 0.1 g crude crystalline 1 from Fr. 17 eluted by CHCl₃-MeOH (24:1). The crude crystals were purified by crystallization (hexane-Me₂CO 1:1) to give 0.04 g 1 and 2.8 g 2.

Maoecrystal I (1). Mp. 205–206°, $C_{22}H_{30}O_8$ (Found: C, 63.20; H, 7.21. $C_{22}H_{30}O_8$ requires: C, 62.56; H, 7.11%), IR $\nu_{\rm max}^{\rm Nujol}$ cm⁻¹: 3200, 2950, 1739, 1705, 1641, 1460, 1240; UV $\lambda_{\rm max}^{\rm McOH}$ nm (log ε) 240.2 (3.6); ¹H and ¹³C NMR spectra: see Tables 1 and 2; EIMS 70 eV, m/z (rel. int.): 422 [M]⁺ (100), 404 [M – H₂O]⁺ (40), 362 [M–AcOH]⁺ (80), 344 [M – AcOH–H₂O]⁺ (55) and 326 [M –

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AcOH $-2 \times H_2O$]⁺ (30); CD curve (MeOH) $[\theta]_{243} - 10761$, $[\theta]_{338} - 2532$.

Maoecrystal J (2). Mp. 249–250°, [α]_D = -49.2 (MeOH; c 1.0); C₂₄H₃₂O₈ (Found: C, 64.15; H, 7.21. C₂₄H₃₂O₈ requires: C, 64.28; H, 7.14%); IR $v_{\rm max}^{\rm Nujol}$ cm⁻¹: 3760, 3360, 2940, 1741, 1708, 1640, 1460, 1239; UV_{max}^{MeOH} nm (log ε) 239.6 (3.9); ¹H and ¹³C NMR spectra: see Tables 1 and 2; EIMS 70 eV m/z (rel. int.): 448 [M]⁺ (40), 430 [M - H₂O]⁺ (27), 388 [M - AcOH]⁺ (100), 328 [M - 2 × AcOH]⁺ (55) and 310 [M - 2 × AcOH - H₂O]⁺ (30); CD curve (MeOH) $\{\theta\}_{240}$ - 18 816, $\{\theta\}_{338}$ - 4928.

Diacetate of 2 (4) and monoacetate of 2 (5). 1.5 mg of 2 was acetylated with Ac_2O in pyridine at room temp. overnight, then the reaction mixture was poured into ice- H_2O and absorbed to SEP-PAK C_{18} . The column was eluted with MeOH, and the crude product was purified by HPLC using ODS₁₈ and 50% MeOH to give 5 (800 μ g) and 6 (500 μ g).

4, FDMS m/z: 507 [M + H]⁺; $\delta_{\rm H}$ (CD₃OD, 400 MHz): 4.60 (1-H, t, 3.0), 2.20 and 2.07 (2-H₂, each 1H, ddd, 16.0, 3.0, 3.0), 5.11 (3-H, t, 3.0), 2.26 and 3.94 (5-H and 6-H, each 1H, d, 5.0), 2.06 (9-H, ddd, 12.0, 7.0, 1.5), 1.89 and 1.82 (11-H₂, each 1H, m), 2.40 (12-H, ddd, 13.5, 9.0, 9.0), 1.50 (12-H, ddd, 13.5, 13.0, 7.0), 3.18 (13-H, overlap), 2.26 (14-H, d, 12.0), 2.19 (14-H, dd, 12.0, 5.0), 6.00 and 5.55 (17-H₂, each 1-H, br s), 1.21 (18-CH₃, s), 4.50 and 4.43 (19-H₂, each 1H, d, 11.5), 4.18 (20-H, dd, 11.0, 1.5), 3.81 (20-H, dd, 11.0, 2.5) as well as 3 × AcO: 2.09, 2.11, 2.12 (each 3H, s).

5, FDMS: m/z 465 [M + H] $^+$; $\delta_{\rm H}$ (CD $_3$ OD, 400 MHz): 3.48 (1-H, t, 3.0), 2.08 and 2.02 (2-H $_2$, each 1H, ddd, 16.0, 3.0, 3.0), 5.06 (3-H, t, 3.0), 2.13 and 3.87 (5-H and 6-H, each 1H, d, 5.0), 2.00 (9-H, br d, 12.0), 1.92 and 1.86 (11-H $_2$, each 1H, m), 2.38 (12-H, ddd, 13.5, 9.0, 9.0), 1.44 (12-H, ddd, 13.0, 13.0, 7.0), 2.21 (14-H, d, 12.0), 2.17 (14-H, dd, 12.0, 5.0), 5.94 and 5.48 (17-H $_2$, br s), 1.17 (18-Me, s), 4.45 and 4.39 (19-H $_2$, each 1H, d, 11.5), 4.09 (20-H, dd, 11.0, 1.5), 3.70 (20-H, dd, 11.0, 2.5) as well as 2 × AcO: 2.07 and 2.09 (each 3H, s).

Periodate oxidation of 2. 2 (5 mg) was dissolved in MeOH (1 ml) containing HIO₄ acid (35 mg) and stirred for 20 hr at room temp. The reaction soln into which 3 ml $\rm H_2O$ was added was evapd. The reaction product was adsorbed to SEP-PAK $\rm C_{18}$ and eluted with MeOH. The solvent was evapd to give a residue (8 mg) which was purified by HPLC using ODS₁₈ and 50% MeOH to afford 6 (2 mg).

6, FDMS m/z: 447 [M+H]⁺; $\delta_{\rm H}$ (CD₃OD, 400 MHz): 1.52 and 1.80 (H-1 and H-2, overlap), 1.86 and 2.26 (H-1 and H-2, each 1H, br d, 15.0), 4.93 (H-3, s), 2.99 and 9.90 (H-5 and H-6

[CHO], each 1H, d, 5.0), 1.85 and 1.82 (H-9 and H-11, overlap), 1.66 (H-11, m), 2.06 and 1.50 (H-12, each 1H, m), 3.17 (H-13, dd, 9.0, 5.5), 2.34 (H-14, d, 13.0), 2.42 (H-14, dd, 13.0, 5.5), 6.02 and 5.60 (H-17, each 1H, s), 4.18 and 4.22 (H-19, each 1H, s), 4.29 and 4.33 (H-20, each 1H, d, 10.5) as well as $2 \times AcO$: 2.08 and 2.15 (each 3H, s).

Hydrogenation of 2. 2 (12 mg) was hydrogenated over Pt_2O (5 mg) in MeOH (3 ml) at room temp. for 1.5 hr. The reaction mixture was treated as usual to give crude product, which was purified by HPLC using ODS_{18} and 50% MeOH to give 7 (8 mg).

7, FABMS: m/z 451 [M+H]⁺; $\delta_{\rm H}$ (CD₃OD, 400 MHz): 1.21 (1-H, br d, 14.0), 1.64 and 1.66 (1-H and 2-H, each 1H, overlap), 1.74 (2-H, br d, 15.0), 5.02 (3-H, t, 2.5), 1.83 (5-H, d, 5.0), 3.75 (6-H, d, 5.0), 1.45 (9-H, dd, 14.0, 5.0), 1.65 and 1.86 (11-H₂, each 1H, overlap), 1.35 and 1.43 (12-H₂, each 1H, overlap), 2.55 (13-H, m), 2.20 (14-H_a, d, 12.0), 2.10 (14-H_b, dd, 12.0, 5.0), 2.45 (16-H, quintet, 6.0), 1.08 (17-Me, d, 6.0), 1.13 (18-Me, s), 4.43 and 4.36 (19-H₂, each 1H, 11.5), 3.97 and 3.86 (20-H₂, each 1H, 10.5) as well as 2.08 (6H, s, 2 × AcO).

Bioassays. The root growth inhibitory activities of 1 and 2 were carried out using lettuce seedlings in Hoagland aq. soln. The test doses used were 1, 2, 5, 10, 50, 100, 200, 500 and 1000 ppm. 10 seeds were used in each test.

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